



Chemistry in Everyday Life

Q.No.1: A water sample has ppm level concentration of following anions

$F^- = 10$; $SO_4^{2-} = 100$; $NO_3^- = 50$

The anion/anions that make/makes the water sample unsuitable for drinking is/are :

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- A. both SO_4^{2-} and NO_3^-
- B. only F^-
- C. only SO_4^{2-}
- D. only NO_3^-

Q.No.2: The correct match between Item-I and Item-II is:

Item-I (drug)	Item-II (test)
A Chloroxylenol	P Carbylamine test
B Norethindrone	Q Sodium hydrogen-carbonate test
C Sulphapyridine	R Ferric chloride test
D Penicillin	S Bayer's test

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- A. $A \rightarrow R$; $B \rightarrow P$; $C \rightarrow S$; $D \rightarrow Q$
- B. $A \rightarrow Q$; $B \rightarrow S$; $C \rightarrow P$; $D \rightarrow R$
- C. $A \rightarrow R$; $B \rightarrow S$; $C \rightarrow P$; $D \rightarrow Q$
- D. $A \rightarrow Q$; $B \rightarrow P$; $C \rightarrow S$; $D \rightarrow R$

Q.No.3: A water sample has ppm level concentration of the following metals:

$Fe = 0.2$; $Mn = 5.0$; $Cu = 3.0$; $Zn = 5.0$. The metal that makes the water sample unsuitable for drinking is:

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- A. Cu
- B. Mn
- C. Fe

D. Zn

Q.No.4: Which of the following conditions in drinking water causes methemoglobinemia?

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- A.** > 50 ppm of lead
- B.** > 50 ppm of chloride
- C.** > 50 ppm of nitrate
- D.** > 100 ppm of sulphate

Q.No.5: The pH of rain water, is approximately:

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- A.** 5.6
- B.** 7.5
- C.** 7.0
- D.** 6.5

Q.No.6: Water filled in two glasses A and B have BOD values of 10 and 20, respectively. Which of the following is the correct statement?

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- A.** B is more polluted than A.
- B.** A is suitable for drinking, whereas B is not.
- C.** Both A and B are suitable for drinking.
- D.** A is more polluted than B.

Q.No.7: The reaction that is **NOT** involved in the ozone layer depletion mechanism in the stratosphere is :

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- A.** $\text{CF}_2\text{Cl}_2(\text{g}) \xrightarrow{\text{uv}} \dot{\text{Cl}}(\text{g}) + \dot{\text{CF}}_2\text{Cl}(\text{g})$
- B.** $\text{Cl}\dot{\text{O}}(\text{g}) + \text{O}(\text{g}) \rightarrow \dot{\text{Cl}}(\text{g}) + \text{O}_2(\text{g})$
- C.** $\text{CH}_4 + 2\text{O}_3 \rightarrow 3\text{CH}_2 = \text{O} + 3\text{H}_2\text{O}$
- D.** $\text{HOCl}(\text{g}) \xrightarrow{\text{hv}} \dot{\text{O}}\text{H}(\text{g}) + \dot{\text{Cl}}(\text{g})$

Q.No.8: The correct match between **Item I** and **Item II** is :

- | Item I | Item II |
|-----------------------|---|
| (A) Allosteric effect | (P) Molecule binding to the active site of enzyme |

- (B) Competitive inhibitor (Q) Molecule crucial for communication in the body
 (C) Receptor (R) Molecule binding to a site other than the active site of enzyme
 (D) Poison (S) Molecule binding to the enzyme covalently

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- A.** (A) → (R); (B) → (P); (C) → (Q); (D) → (S)
B. (A) → (P); (B) → (R); (C) → (Q); (D) → (S)
C. (A) → (R); (B) → (P); (C) → (S); (D) → (Q)
D. (A) → (P); (B) → (R); (C) → (S); (D) → (Q)

Q.No.9: The correct match between item (I) and item (II) is:

Item - I	Item - II
(A) Norethindrone	(P) Anti-biotic
(B) Ofloxacin	(Q) Anti-fertility
(C) Equanil	(R) Hypertension
	(S) Analgesics

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- A.** (A) → (Q); (B) → (R); (C) → (S)
B. (A) → (Q); (B) → (P); (C) → (R)
C. (A) → (R); (B) → (P); (C) → (S)
D. (A) → (R); (B) → (P); (C) → (R)

Q.No.10: The concentration of dissolved oxygen (DO) in cold water can go upto:

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- A.** 14 ppm
B. 8 ppm
C. 10 ppm
D. 16 ppm