

Haloalkanes and Haloarenes

Q.No.1: The major organic compound formed by the reaction of 1, 1, 1 - trichloroethane with silver powder is

- A. 2 Butyne
- **B.** 2 Butene
- C. Acetylene
- **D.** Ethene

Q.No.2:

2-Chloro-2-methylpentane on reaction with sodium methoxide in methanol yields:

- (a) CH₃CH₂CH₂C(CH₃)₂OCH₃ (b) CH₃CH₂CH₂C(CH₃)CH₂
- (c) CH₃CH₂CHC(CH₃)₂

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- **A.** (a) and (c)
- **B.** (c) only
- **C.** (a) and (b)
- D. All of these

Q.No.3: The absolute configuration of

CO₂H H — OH H — Cl CH₃ ;

is

- **A.** (2S, 3R)
- **B.** (2S, 3S)
- **C.** (2R, 3R)
- **D.** (2R, 3S)

Q.No.4: The reaction of propene with HOCl ($Cl_2 + H_2O$) proceeds through the intermediate: **JEE 2016**

A.
$$CH_3 - CH^+ - CH_2 - CI$$

B.
$$CH_3 - CH(OH) - CH_2^+$$

C.
$$CH_3$$
 - $CHCI$ - CH_2 +

D.
$$CH_3 - CH^+ - CH_2 - OH$$

Q.No.5: The increasing order of the reactivity of the following halides for the $S_N {\bf 1}$ reaction is :

- **A.** (II) < (I) < (III)
- **B.** (I) < (III) < (II)
- **C.** (II) < (III) < (I)
- **D.** (III) < (II) < (I)

Q.No.6: The compounds A and B in the following reaction are, respectively:

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- **A.** A = Benzyl alcohol, B = Benzyl cyanide
- **B.** A = Benzyl chloride, B = Benzyl cyanide
- C. A = Benzyl alcohol, B = Benzyl isocyanide
- **D.** A = Benzyl chloride, B = Benzyl isocyanide

Q.No.7: The major product of the following reaction is:

В. OCH_3

C. CH₃O

D. CH₃O

Q.No.8: Identify A and B in the chemical reaction.

 OCH_3

A.

В.

D.

Q.No.9: Identify A in the given reaction,

A.

Cl

Q.No.10: Match List-I with List-II

CH₂Cl

| List-I | List-II |
|--|----------------------------|
| $\begin{array}{c c} & C1 \\ & Cu_2Cl_2 \\ & \end{array} + N_2 $ | (i) Wurtz reaction |
| b) $N_2^+Cl^ Cu/HCl$ $+ N_2$ | (ii) Sandmeyer reaction |
| c) $	ext{2 CH}_3 	ext{CH}_2 	ext{Cl} + 	ext{2 Na} \stackrel{	ext{Ether}}{\longrightarrow} 	ext{C}_2 	ext{H}_5 - 	ext{C}_2 	ext{H}_5 + 	ext{2 NaCl}$ | (iii) Fittig reaction |
| d) $2	ext{C}_6	ext{H}_5	ext{Cl} \ + \ 2	ext{Na} \ \stackrel{	ext{Ether}}{\longrightarrow} \ 	ext{C}_6	ext{H}_5 - 	ext{C}_6	ext{H}_5 \ + \ 2	ext{NaCl}$ | (iv) Gatterman reaction |

Choose the correct answer from the options given below

- **A.** (a)-(iii); (b)-(iv); (c)-(i); (d)-(ii)
- **B.** (a)-(ii); (b)-(iv); (c)-(i); (d)-(iii)
- **C.** (a)-(iii); (b)-(i); (c)-(iv); (d)-(ii)
- **D.** (a)-(ii); (b)-(i); (c)-(iv); (d)-(iii)