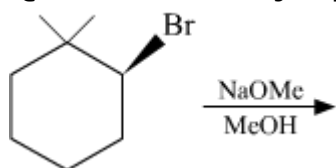




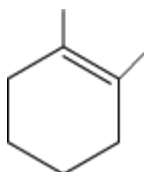
Organic Chemistry: Some Basic Principles and Techniques

Q.No.1: The major product of the following reaction is :

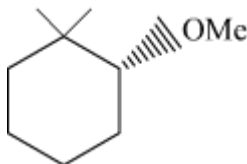
JEE 2018



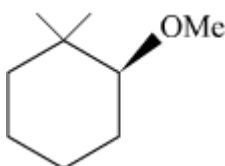
A.



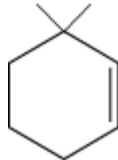
B.



C.



D.



Q.No.2: An aqueous solution contains an unknown concentration of Ba^{2+} . When 50 mL of a 1 M solution of Na_2SO_4 is added, BaSO_4 just begins to precipitate. The final volume is 500 mL. The solubility product of BaSO_4 is 1×10^{-10} . What is the original concentration of Ba^{2+} ?

JEE 2018

A. $1.1 \times 10^{-9} \text{ M}$

B. $1.0 \times 10^{-10} \text{ M}$

C. $5 \times 10^{-9} \text{ M}$

D. 2×10^{-9} M

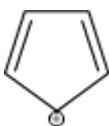
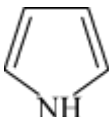
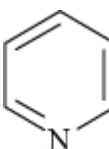
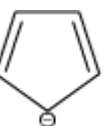
Q.No.3: Which amongst the following is the strongest acid?

JEE 2019

- A. CHBr_3
- B. CHI_3
- C. $\text{CH}(\text{CN})_3$
- D. CHCl_3

Q.No.4: Which of the following compounds is not aromatic?

JEE 2019

- A. A five-membered ring with two double bonds and a negative charge at the bottom vertex.
- B. A five-membered aromatic heterocycle with one nitrogen atom at the bottom vertex.
- C. A six-membered aromatic heterocycle with one nitrogen atom at the bottom vertex.
- D. A five-membered ring with two double bonds and a positive charge at the bottom vertex.

Q.No.5: If dichloromethane (DCM) and water (H_2O) are used for differential extraction, which one of the following statements is correct?

JEE 2019

- A. DCM and H_2O would stay as lower and upper layer, respectively in the separating funnel (S.F.).
- B. DCM and H_2O will make turbid/colloidal mixture.
- C. DCM and H_2O would stay as upper and lower layer, respectively in the separating funnel (S.F.).
- D. DCM and H_2O will be miscible clearly.

Q.No.6: The correct match between item 'I' and item 'II' is:

Item 'I' (compound)	Item 'II' (compound)
(A) Lysine	(P) 1-naphthol

- (B) Furfural
(C) Benzyl alcohol
(D) Styrene
- (Q) ninhydrin
(R) KMnO_4
(S) Ceric ammonium nitrate

JEE 2019

- A.** (A) \rightarrow (Q); (B) \rightarrow (P); (C) \rightarrow (S); (D) \rightarrow (R)
B. (A) \rightarrow (Q); (B) \rightarrow (P); (C) \rightarrow (R); (D) \rightarrow (S)
C. (A) \rightarrow (R); (B) \rightarrow (P); (C) \rightarrow (Q); (D) \rightarrow (S)
D. (A) \rightarrow (Q); (B) \rightarrow (R); (C) \rightarrow (S); (D) \rightarrow (P)

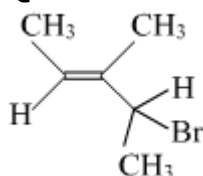
Q.No.7: Sodium metal on dissolution in liquid ammonia gives a deep blue solution due to the formation of :

JEE 2019

- A.** sodium-ammonia complex
B. sodamide
C. sodium ion-ammonia complex
D. ammoniated electrons

Q.No.8: What is the IUPAC name of the following compound?

JEE 2019



- A.** 3-Bromo-1, 2-dimethylbut-1-ene
B. 3-Bromo-3-methyl-1, 2-dimethylprop-1-ene
C. 2-Bromo-3-methylpent-3-ene
D. 4-Bromo-3-methylpent-2-ene

Q.No.9: The correct match between items I and II is:

Item - I (Mixture)	Item - II (Separation method)
(A) H_2O : Sugar	(P) Sublimation
(B) H_2O : Aniline	(Q) Recrystallization
(C) H_2O : Toluene	(R) Steam distillation
	(S) Differential extraction

JEE 2019

- A.** (A) \rightarrow (S); (B) \rightarrow (R); (C) \rightarrow (P)
B. (A) \rightarrow (Q); (B) \rightarrow (R); (C) \rightarrow (S)
C. (A) \rightarrow (R); (B) \rightarrow (P); (C) \rightarrow (S)
D. (A) \rightarrow (Q); (B) \rightarrow (R); (C) \rightarrow (P)

Q.No.10: Which of the following compounds will produce a precipitate with AgNO_3 ?

JEE 2019

